

NAMING COMPOUNDS (II)

- |                           |                            |
|---------------------------|----------------------------|
| 1. $\text{CO}_2$          | 8. $\text{P}_3\text{N}_5$  |
| 2. $\text{SO}_3$          | 9. $\text{SbS}_5$          |
| 3. $\text{N}_2\text{O}$   | 10. $\text{PBr}_5$         |
| 4. $\text{CO}$            | 11. $\text{SiO}_2$         |
| 5. $\text{P}_2\text{O}_5$ | 12. $\text{N}_2\text{O}_5$ |
| 6. $\text{N}_2\text{O}_3$ | 13. $\text{SO}_2$          |
| 7. $\text{NO}_2$          | 14. $\text{NO}$            |

WRITING FORMULAS (II)

- |                           |                             |
|---------------------------|-----------------------------|
| 1. manganese dioxide      | 7. carbon dioxide           |
| 2. sulfur dioxide         | 8. dinitrogen pentoxide     |
| 3. diphosphorus pentoxide | 9. antimony pentasulfide    |
| 4. carbon monoxide        | 10. selenium tetrafluoride  |
| 5. nitrogen dioxide       | 11. phosphorus pentabromide |
| 6. phosphorus trichloride | 12. silicon dioxide         |